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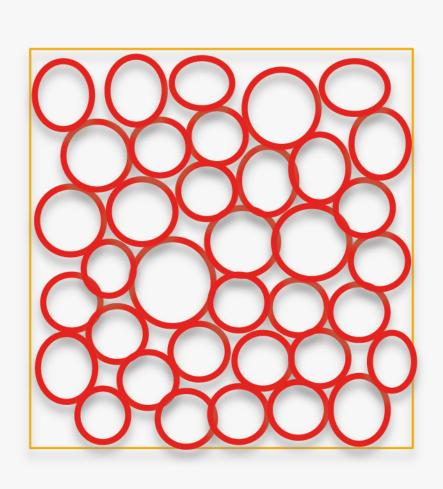
The size of droplets can play an important role when it comes to texture and stability of an emulsion. Especially during the scalling-up of laboratory batches for production, different droplet sizes often occur.

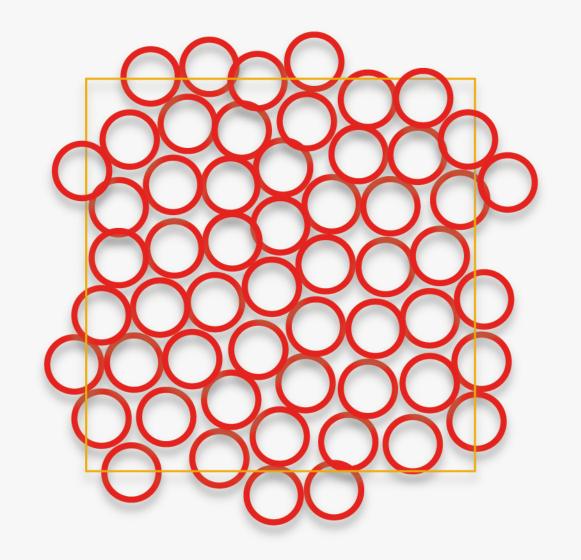


If the internal phase of the production batch is dispersed more finely (= smaller droplets), there is a risk of an increase in viscosity. A bodylotion for example could turn into a cream or even a butter-like consistency. In this case it is also very difficult to apply the product on the finger.



The cause for this "oil migration" is actually a space problem. In such cases the oil phase does not have enough space and is therefore (at least partially) pushed out of the system.









- The internal phase has reached an unfavourable size (too large) in relation to the external phase.
- Gelling agents in the external phase bind a lot of space due to their swelling behaviour.
- The smaller the droplets in the internal phase are, the more space is taken up.



- A strong gel network exists in the water phase (like carbomers, xanthan gum, etc.)
- The oil phase is relatively large and therefore demands a certain amount of space within the emulsion.
- The finer the droplets become, the more space the oil phase requires.

